

U.S. DEPARTMENT OF COMMERCE, PATENT AND TRADEMARK OFFICE		DATE: August 27, 2001
TRANSMITTAL LETTER TO THE UNITED STATES DESIGNATED/ELECTED OFFICE (DO/EO/US) CONCERNING A FILING UNDER 35 U.S.C. 371		U.S. APPL. NO. (if known) 09/926085 ✓
INTERNATIONAL APPLICATION NO.: PCT/JP00/02435 ✓	INTERNATIONAL FILING DATE: April 14, 2000 ✓	PRIORITY DATE CLAIMED: April 23, 1999 ✓
TITLE OF INVENTION: PROCESSING AID FOR THERMOPLASTIC RESIN AND THERMOPLASTIC RESIN COMPOSITION CONTAINING THE SAME ✓		
APPLICANT(S) FOR DO/EO/US: Takenobu SUNAGAWA, Riichi NISHIMURA, Toshiyuki MORI and Akira TAKAKI ✓		
Applicant hereby submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:		
<p>1. <u>XX</u> This is a <b>FIRST</b> submission of items concerning a filing under 35 U.S.C. 371.</p> <p>2. <u>  </u> This is a <b>SECOND</b> or <b>SUBSEQUENT</b> submission of items concerning a filing under 35 U.S.C. 371.</p> <p>3. <u>XX</u> This express request to begin national examination procedures (35 USC 371(f)) at any time rather than delay examination until the expiration of the time limit set in 35 USC 371(b) and PCT Articles 22 and 39(1).</p> <p>4. <u>XX</u> A proper Demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date.</p> <p>5. <u>XX</u> A copy of the International Application as filed (35 U.S.C. 371(c)(2)):</p> <p style="margin-left: 20px;">a. <u>  </u> is transmitted herewith (required only if not transmitted by the International Bureau).</p> <p style="margin-left: 20px;">b. <u>XX</u> has been transmitted by the International Bureau.</p> <p style="margin-left: 20px;">c. <u>  </u> is not required, as the application was filed in the United States Receiving Office (RO/US).</p> <p>6. <u>XX</u> A translation of the International Application into English (35 U.S.C. 371(c)(2)).</p> <p>7. <u>XX</u> Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371(c)(3))</p> <p style="margin-left: 20px;">a. <u>  </u> are transmitted herewith (required only if not transmitted by the International Bureau).</p> <p style="margin-left: 20px;">b. <u>  </u> have been transmitted by the International Bureau.</p> <p style="margin-left: 20px;">c. <u>  </u> have not been made; however, the time limit for making such amendments has NOT expired.</p> <p style="margin-left: 20px;">d. <u>XX</u> have not been made and will not be made.</p> <p>8. <u>  </u> A translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).</p> <p>9. <u>XX</u> An oath or declaration of the inventor(s) (35 U.S.C. 371(c)(4)).</p> <p>10. <u>  </u> A translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371(c)(5)).</p> <p><b>ITEMS 11. TO 16. BELOW CONCERN OTHER DOCUMENT(S) OR INFORMATION INCLUDED:</b></p> <p>11. <u>XX</u> An Information Disclosure Statement under 37 CFR 1.97 and 1.98 together with the <u>International Search Report</u> and 5 references</p> <p>12. <u>XX</u> An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included. ASSIGNEE NAME AND ADDRESS: <b>KANEKA CORPORATION, Osaka-shi, Japan</b></p> <p style="text-align: center;"><b>Please publish the assignee data with the application.</b></p> <p>13. <u>  </u> A <b>FIRST</b> preliminary amendment.       <u>  </u> A <b>SECOND</b> or <b>SUBSEQUENT</b> preliminary amendment</p> <p>14. <u>  </u> A substitute specification.</p> <p>15. <u>  </u> A change of power of attorney and/or address letter.</p> <p>16. <u>  </u> Other items or information:</p>		

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17. <input checked="" type="checkbox"/> The following fees are submitted:  <b>Basic National Fee (37 CFR 1.492(a)(1)-(5):</b> Search Report has been prepared by the EPO or JPO: ..... \$860.00  International preliminary examination fee paid to USPTO (37 CFR 1.482) ..... \$690.00  No international preliminary examination fee paid to USPTO (37 CFR 1.482) but international search fee paid to USPTO (37 CFR 1.445(a)(2)) ..... \$710.00  Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO ..... \$1000.00  International preliminary examination fee (37 CFR 1.482) and all claims satisfied provisions of PCT Article 33(2)-(4) ..... \$100.00  <div style="text-align: right;"><b>ENTER APPROPRIATE BASIC FEE AMOUNT = \$ 860.00</b></div>	CALCULATIONS	PTO USE ONLY
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Surcharge of <b>\$130.00</b> for furnishing the oath or declaration later than <u>XX</u> 20 30 months from the earliest claimed priority date (37 CFR 1.492(e)).		
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CLAIMS	NUMBER FILED	NUMBER EXTRA	RATE		
TOTAL	8 -20 =		X \$ 18.00		
INDEPENDENT	1 - 3 =		X \$ 80.00		
Multiple dependent claims(s) (if applicable)			+ \$270.00		
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<b>SUBTOTAL =</b>				<b>\$860.00</b>	
Processing fee of <b>\$130.00</b> for furnishing the English translation later than <u>   </u> 20 30 months from the earliest claimed priority date (37 CFR 1.492(f)).				+	
<b>TOTAL NATIONAL FEE =</b>				<b>\$860.00</b>	
Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31). <b>\$40.00 per property +</b>				<b>\$40.00</b>	
<b>TOTAL FEES ENCLOSED =</b>				<b>\$ 900.00</b>	
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				charged	\$ _____

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a. ☒ A check in the amount of **\$900.00** to cover the above fees is enclosed. (\$860.00 for basic filing fee and \$40.00 for assignment recordation fee).


b. ☐ Please charge my Deposit Account No. 01-2340 in the amount of \$     to cover the above fees. (A duplicate copy of this sheet is enclosed.)

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**NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed to request that the application be restored to pending status.**

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## DESCRIPTION

PROCESSING AID FOR THERMOPLASTIC RESIN AND THERMOPLASTIC  
RESIN COMPOSITION CONTAINING THE SAME

5

## TECHNICAL FIELD

The present invention relates to a processing aid for thermoplastic resin molding such as calendar molding, blow molding, extrusion molding and injection molding, and a thermoplastic resin composition containing the processing aid, which is excellent in processability, particularly, in peeling property from a metal surface having a high temperature.

## BACKGROUND ART

Thermoplastic resins, such as vinylchloride resins, polyethylene, polypropylene, polyester, polystyrene, polycarbonate, a methacrylate resin, and ABS resin, are excellent in physical properties, and have economic advantage and applicability, and have been widely used in various fields. However, since these resins have high melt viscosity and low fluidity, and are susceptible to thermal decomposition, there are problems in processability that they have a narrow processable condition in molding, and that they tend to stick or adhere to a metal surface of a device in processing at a high temperature.

Now there have been proposed many techniques in order to solve the above-mentioned problems in processability.

Applicability of vinylchloride resins, polyethylene, polypropylene, polyester, polystyrene, polycarbonate, a methacrylate

resin, ABS resin and the like can be improved by physically blending modifiers such as a lubricant, a stabilizer, a colorant, a filler, a pigment, a cross-linking agent, a tackifier, a plasticizer, a processing aid, an impact modifier and a modifier for thermal deformation temperature,  
5 before processing.

Moreover, in order to improve molding processability at molding of vinylchloride resins, polyethylene, polypropylene, polyester, polystyrene, polycarbonate, a methacrylate resin, and ABS resin, some of copolymers compatible with those resins have been investigated as a  
10 processing aid.

For example, there have been proposed a method in which a copolymer of methyl methacrylate and styrene is added to vinylchloride resins (Japanese Examined Patent Publication No. 32-4140 (1957)), a method in which a copolymer of styrene and acrylonitrile is added  
15 thereto (Japanese Examined Patent Publication No. 29-5246 (1954)), a method in which a copolymer mainly containing methyl methacrylate is added thereto (Japanese Examined Patent Publication No. 40-5311 (1965)), a method in which a copolymer of styrene and alkyl acrylate is added thereto (Japanese Examined Patent Publication No. 37-1384  
20 (1962)) and the like. Secondary processability such as an increase in melting rate of vinyl chloride resins and an increase in tensile elongation at a high temperature can be improved, but there is no effect in reducing adhesiveness to a metal face of the molding device.

Thereafter, in order to reduce adhesiveness to a metal surface  
25 of a molding device, there has been investigated a method in which a copolymer of styrene, an alkyl (meth)acrylate and (meth)acrylate having an oxygen atom in addition to an ester bond is added to vinyl chloride

resins (Japanese Examined Patent Publication No. 58-56536 (1983)). In this method the adhesiveness is proved to be improved, but this method cannot always satisfy the demands in the market.

#### DISCLOSURE OF THE INVENTION

Taking the above-mentioned circumstances into consideration, it has been found that the above-mentioned problems could be solved by using a polymer as a processing aid, which was obtained by polymerizing a specific amount of monomers having a specific composition and a specific chain transfer agent to complete the present invention. Moreover, it has been found out that the above-mentioned problems could be solved by using a polymer as a processing aid, which was obtained by polymerizing a specific amount of monomers having a specific composition and a specific polymerization initiator to complete the present invention.

Namely, the present invention relates to a processing aid for thermoplastic resin having a weight average molecular weight of 10,000 to 300,000, which is obtained by polymerizing an alkyl (meth)acrylate, or an alkyl (meth)acrylate and another vinyl monomer copolymerizable therewith, in the presence of a mercaptan having an alkyl ester group with 4 to 20 of alkyl carbon as a chain transfer agent, and/or an organic peroxide having a tertiary-butyl peroxy group as a polymerization initiator.

#### BEST MODE FOR CARRYING OUT THE INVENTION

The present invention features that a copolymer having a weight average molecular weight of 10,000 to 300,000 is used as a

processing aid for thermoplastic resin, which is obtained by polymerizing an alkyl (meth)acrylate, or a monomer mixture comprising an alkyl (meth)acrylate and another vinyl monomer copolymerizable therewith, in the presence of a mercaptan having an alkyl ester group with 4 to 20 of alkyl carbon as a chain transfer agent, and/or an organic peroxide having a tertiary-butyl peroxy group as a polymerization initiator. More preferably, the present invention features that a copolymer is used as a processing aid for thermoplastic resin, which is obtained by polymerizing a mixture comprising a (meth)acrylate having an oxygen atom in addition to an ester bond, another alkyl acrylate, another vinyl monomer copolymerizable therewith, in the presence of a mercaptan having an alkyl ester group with 4 to 20 of alkyl carbon as an organic peroxide having a tertiary-butylperoxy group as a polymerization initiator.

By using the copolymer as the processing aid in an amount of 0.1 to 20 parts by weight based on the thermoplastic resin, processability thereof can be improved without losing physical or chemical properties that the resin originally has. Particularly, there arise effects expected by adding the above modifier, such as drastically improving removability from a metal surface at a high temperature.

The processing aid of the present invention, comprising a copolymer obtained by emulsion polymerizing the above-mentioned monomer mixture, can provide excellent processability, in particular, properties such as a peeling property from a metal surface at a high temperature without degradation in physical and chemical properties of a thermoplastic resin.

The above-mentioned monomer mixture is a mixture

comprising (meth)acrylate having an oxygen atom in addition to an ester bond in an amount of preferably 0.1 to 10 % by weight, more preferably 0.5 to 5 % by weight, and particularly preferably 0.5 to 2 % by weight; another alkyl acrylate in an amount of preferably 10 to 99.9 % by weight, more preferably 10 to 59.5 % by weight, particularly preferably, 10 to 39.5 % by weight; and another vinyl monomer copolymerizable therewith in an amount of preferably 0 to 89.9 % by weight, more preferably 40 to 89.5 % by weight, and particularly preferably 60 to 89.5 % by weight.

Examples of the (meth)acrylate having an oxygen atom in addition to an ester bond are acrylate containing an epoxy group such as glycidyl acrylate; methacrylate containing an epoxy group such as glycidyl methacrylate; acrylate containing a hydroxy group such as 2-hydroxyethylacrylate and 2-hydroxypropylacrylate; methacrylate containing a hydroxy group such as 2-hydroxyethylmethacrylate and 2-hydroxypropylmethacrylate; acrylate containing an alkoxy group such as methoxyethylacrylate and ethoxyethylacrylate; methacrylate containing an alkoxy group such as methoxyethylmethacrylate and ethoxyethylmethacrylate, and the like. These materials may be used solely or in a combination use of two or more thereof. In order to obtain an excellent peeling property from a metal surface at a high temperature, these materials preferably comprises 0.1 to 10 % by weight. If the content is less than or more than this range, a sufficient peeling effect can not be expected.

Another alkyl acrylate is preferably an alkyl acrylate containing an alkyl group having 1 to 20 carbon atoms. Examples thereof are those having an alkyl group having 3 to 8 carbon atoms such as 2-ethylhexylacrylate, butylacrylate, and ethylacrylate. These may be

used solely or in a combination use of two or more thereof. In order to obtain an excellent peeling property from a metal surface at a high temperature, these materials preferably comprise 10 to 99.9 % by weight thereof. If the content is less or more than this range, a sufficient  
5 peeling effect can not be expected.

Examples of the other vinyl monomer that is copolymerizable with a small amount of (meth)acrylate containing an oxygen atom in addition to an ester bond and the other alkyl acrylate are alkyl methacrylates such as methyl methacrylate and butyl methacrylate,  
10 aromatic vinyls such as styrene,  $\alpha$ -methylstyrene and chlorostyrene, acrylonitrile, methacrylonitrile. These may be used solely or in a combination use of two or more thereof. In order to obtain an excellent peeling property from a metal surface at a high temperature, these materials preferably comprises 0 to 89.9 weight %. If the content is less  
15 or more than this range, a sufficient peeling effect can not be expected.

There is no particular limitation for the chain transfer agent. Conventional chain transfer agents, such as tertiary- dodecyl mercaptan, n-dodecyl mercaptan, tertiary-decyl mercaptan and n-decyl mercaptan, may be used depending on demand. An amount of the chain transfer  
20 agent is preferably 0.5 to 3 parts by weight based on 100 parts by weight of the monomers in total.

In particular, a preferable chain transfer agent is a mercaptan containing an alkyl ester group having C<sub>4-20</sub> alkyl group. Examples thereof are tertiary-butylthioglycolate, 2-ethylhexylthioglycolate, 2-  
25 ethylhexyl-  $\beta$  -mercaptopropionate, and tridecylmercaptopropionate. These may be used solely or in a combination use of two or more thereof. In order to obtain an excellent peeling property from a metal surface at a

high temperature, these materials comprise preferably 0.5 to 3 parts by weight based on 100 parts by weight of the monomers in total.

From the viewpoint of improving the peeling property from a metal surface at a high temperature, a weight average molecular weight of the copolymer comprising the monomer mixture is preferably 10,000 to 300,000.

There is no particular limitation for the polymerization initiator, but an organic peroxide having a tertiary-butyl peroxy group is preferably used.

10        Examples of the organic peroxide having a tertiary-butyl peroxy group are tertiary-butylhydroperoxide, di(tertiary-butyl)peroxyde, tertiary-butyl-  $\alpha$  -cumylperoxide, tertiary-butylisopropylcarbonate, tertiary-butylperoxyacetate, tertiary-butylperoxyisobutylate, tertiary-butylperoxyoctanate, tertiary-butylperoxylaurate, tertiary-butylperoxypivalate, tertiary-butylperoxyneodecanoate, and tertiary-butylperoxybenzoate. These may be used solely or in a combination use of two or more thereof. In order to obtain an excellent peeling property from a metal surface at a high temperature, an amount thereof is preferably 0.1 to 5 parts by weight, and more preferably 0.5 to 3 parts by weight.

20        The processing aid of the present invention can be prepared by using, for example, the following method. First, the monomer mixture is emulsion polymerized in the presence of a suitable medium, an emulsifier and the chain transfer agent and/or the polymerization initiator. The medium used in the emulsification polymerization is usually water.

The processing aid of the present invention may be a polymer

prepared by single step polymerization or a polymer prepared by a multi-steps polymerization such as two steps or three steps.

As the emulsifier, there can be used a known emulsifier. Examples thereof are anionic surfactants such as a fatty acid salt, an alkyl sulfate, an alkyl benzene sulfate, an alkyl phosphate, and a diester of sulfosuccinic acid, and non-ionic surfactants such as a polyoxyethylene alkyl ether and a polyoxyethylene fatty acid ester.

As the above-mentioned chain transfer agent, there can be used the above-mentioned mercaptan having an alkyl ester group with C<sub>4-20</sub> alkyl group.

As the polymerization initiator, there is usually used a polymerization initiator of the redox type, which contains the above-mentioned organic peroxide having a tertiary-butylperoxy group in combination with a reducing agent. It is possible to use a general organic peroxide in a combination therewith, if necessary.

There is no particular limitation for a temperature, time and the like at the polymerization condition. Usual temperature and time may be selected, and these may be suitably adjusted to obtain a desired molecular weight and particle size.

The processing aid of the present invention is mixed with a thermoplastic resin according to a usual method to obtain the thermoplastic resin composition of the present invention.

An amount at mixing of the thermoplastic resin and the processing aid can be widely set, but the amount thereof is preferably 0.1 to 20 parts by weight, and more preferably 0.5 to 3 parts by weight based on 100 parts by weight of the thermoplastic resin. If the amount is less than 0.1 parts by weight, a sufficient effect can not be obtained.

And if the amount is more than 20 parts by weight, transparency of the molded article becomes low and many fish eyes tend to occur.

The thermoplastic resin includes all normal thermoplastic resins. In particular, there can preferably be vinyl chloride resins, polyethylene, polypropylene, polyester, polystyrene, polycarbonate, a methacrylate resin, ABS resin and the like from the viewpoint of improving the peeling property from a metal surface at a high temperature. Among those, vinyl chloride resins are preferable.

The vinyl chloride resins are a polymer comprising 80 to 100 % by weight of vinyl chloride unit and 0 to 20 % by weight of another monomer copolymerizable with the vinyl chloride monomer. As the other monomer copolymerizable with the vinyl chloride monomer, examples thereof are vinyl acetate, propylene, styrene, acrylate and the like. These may be used solely or in a combination use of two or more thereof.

There is no particular limitation for an average polymerization degree and other factors of the vinyl chloride resins, and there may be employed the conventionally used vinyl chloride resins. Examples of the vinyl chloride resins are poly(vinyl chloride), a copolymer of a vinyl chloride monomer of at least 80 % by weight and another copolymerizable monomer, for example, vinyl acetate, propylene, styrene, acrylate, and the like, and chlorinated poly(vinyl chloride). These include homopolymers, and copolymers which contain at least 80 % by weight of vinyl chloride and chlorinated poly(vinyl chloride). These monomers may be used solely or in a combination use of two or more thereof.

To the vinyl chloride resin composition of the present

invention can be suitably added a stabilizer, a lubricant, an impact modifier, a plasticizer, a colorant, a filler and the like at blending, if necessary.

Example of the above-mentioned polyester is poly(ethylene  
5 terephthalate). Moreover, example of the above-mentioned methacrylate resin is poly(methyl methacrylate).

The thermoplastic resin composition of the present invention is excellent in processability, and they can be molded by a method such as blow molding, injection molding, calendar molding or extrusion  
10 molding. The obtained molded article is excellent in appearance properties such as transparency, glow and surface smoothness, as well as in secondary processability, and the resin composition can be suitably applied for various fields such as a film, a sheet, and a profile molded article.

15 The present invention is explained in detail by using examples and comparative examples. But the present invention is not intended to be limited thereby. Hereinafter, "parts" represents "parts by weight" based on 100 parts by weight of the monomers in total, unless otherwise specified. In examples and comparative examples, BA  
20 represents butyl acrylate, EA represents ethyl acrylate, 2EHA represents 2-ethylhexyl acrylate, St represents styrene, MMA represents methyl methacrylate, AN represents acrylonitrile, GMA represents glycidyl methacrylate, 2HEMA represents 2-hydroxyethyl methacrylate, 2EHTG represents 2-ethylhexyl thioglycolate, ETOM represents ethoxyethyl  
25 methacrylate, GA represents glycidyl acrylate, 2HEA represents 2-hydroxyethyl acrylate, ETOA represents ethoxyethyl acrylate, TBHP represents tertiary- butylhydroperoxide.

The following description shows an evaluation method applied to examples and comparative examples.

(Polymerization conversion)

Polymerization conversion was calculated based on the following equation:

Polymerization conversion (%) = (Amount of produced polymer / Amount of loaded monomer)  $\times$  100(%)

(Roll peeling property)

In order to evaluate the roll peeling property, there was kneaded a mixture comprising 100 parts of poly(vinyl chloride) (Kane vinyl S1007 available from Kaneka Corp.), 1 part of a processing aid, 2.0 parts of a stabilizer of octyl tin mercaptan (TVS #8831 available from Nitto Kasei K.K.), 1.0 part of a lubricant (KALCOHL 8668 available from Kao Corp.), and 3.0 parts of dioctyl phthalate (DOP) by using a 6-inch roll at 190°C, to evaluate peeling property from the roll surface after 10 minutes. The evaluation was made in accordance with ten ranks. Based upon the following criteria, 10 was evaluated as the highest peeling property, and 1 was evaluated as the lowest.

10: A molten sheet maintains a state removable from a hot roll surface for 10 minutes or more.

5: A molten sheet maintains a state removable from a hot roll surface for at least 5 minutes to less than 6 minute.

1: A molten sheet is maintained removable from a hot roll surface for 2 minutes or less.

(Transparency)

In order to evaluate transparency, after the composition was kneaded for 5 minutes at 170°C by using an 8-inch roll, it was then

pressed for 15 minutes at 180°C to prepare a plaque having a thickness of 3 mm. Light transmittance for all light rays and haze of the plaque were measured by using an integral bulb-type light-ray transmittance measuring device according to JIS-6714 standard. It shows that the  
5 larger the value of the transmittance is, the better the transparency becomes, and the smaller the value of the haze, the better the transparency becomes.

#### EXAMPLE 1

10 A reactor equipped with a stirrer and a condenser was charged with 200 parts of distilled water, 1.2 parts of sodium dioctylsulfosuccinate, 0.01 part of sodium ethylenediaminetetraacetate, 0.005 part of ferrous sulfate heptahydrate, and 0.5 parts of sodium formaldehyde hydrosulfite. Then, after inside of the reactor was  
15 replaced by nitrogen, the reactor was heated to 60°C with stirring. Next, to this was continuously added for two hours a mixture comprising 30 % by weight of methyl methacrylate (MMA), 19 % by weight of styrene (St) and 0.2 part of tertiary-butyl hydroperoxide (TBHP). One hour after completion of the addition, to this was further continuously added for  
20 three hours a mixture comprising 35 % by weight of St, 15 % by weight of butyl acrylate (BA), 1.0 % by weight of glycidyl methacrylate (GMA), 1.0 part of 2-ethylhexyl thioglycolate (2 EHTG) and 0.8 parts of TBHP. After completion of the addition, the mixture was further stirred for one hour, and then cooled to obtain a copolymer in latex. Polymerization  
25 conversion was 99.4 %.

The obtained latex was coagulated with an aqueous solution of calcium chloride. After the latex was treated up to 90°C, the latex

was filtrated with a centrifugal dehydrator. The obtained dehydrated cake of the copolymer was washed with water and dried for 15 hours at 50°C with a parallel-flow dryer to obtain a white powdery sample (1) prepared by two steps polymerization. A weight average molecular weight of the obtained sample was 70,000 measured by GPC. The obtained sample was subjected to the above-mentioned evaluation on the roll peeling property. Table 1 shows the results.

#### EXAMPLES 2 to 7 and COMPARATIVE EXAMPLES 1 to 3

According to compositions shown in Table 1, samples (2) to (10) were obtained by the same manner as in Example 1. The obtained samples were subjected to the above-mentioned evaluation on the roll peeling property. Table 1 shows the results.

From the results shown in Table 1, as indicated by samples (1) and (4) to (9), it is found that a composition having an excellent roll peeling property is obtained when a composition of the monomer mixture is in the range of the present invention. On the other hand, there is a reduction in the roll peeling property when the composition is out of the range of the present invention, that is, in the case of samples (2), (3) and (10).

TABLE I

Example No.	Sample	Composition of mixture before polymerization (% by weight)										Chain transfer agent (parts)	Organic peroxide (parts)	Roll peeling property	Weight average molecular weight
		Polymer at first step					Polymer at second step								
		MMA	AN	St	BA	EA	2EHA	St	GMA	2EHTG	TBHP				
Ex. 1	(1)	30		19	15			35	1.0	1.0	1.0			10	70,000
Com. Ex. 1	(2)	30		20	15			35	0	1.0	1.0			4	80,000
Com. Ex. 2	(3)	26		18	13			31	12	1.0	1.0			4	70,000
Ex. 2	(4)	6		4	80			6	1.0	1.0	1.0			9	80,000
Ex. 3	(5)	30		19	11			39	1.0	1.0	1.0			9	70,000
Ex. 4	(6)	30		19		15		35	1.0	1.0	1.0			9	70,000
Ex. 5	(7)	30		19			15	35	1.0	1.0	1.0			9	70,000
Ex. 6	(8)		30	19	15			35	1.0	1.0	1.0			9	80,000
Ex. 7	(9)	49			50				1.0	1.0	1.0			8	70,000
Com. Ex. 3	(10)	30		19	9			41	1.0	1.0	1.0			3	70,000

### EXAMPLE 8

A reactor equipped with a stirrer and a condenser was charged with 200 parts of distilled water, 1.2 parts of sodium dioctylsulfosuccinate, 0.01 part of sodium ethylenediaminetetraacetate, 5 0.005 part of ferrous sulfate heptahydrate, and 0.5 part of sodium formaldehyde hydrosulfite. Then, after inside of the reactor was replaced with nitrogen gas, the reactor was heated to 60°C with stirring. Next, to this was continuously added for two hours a mixture comprising 24 % by weight of MMA, 15 % by weight of St and 0.2 part of TBHP. 10 One hour after completion of the addition, to this was further continuously added for three hours a mixture comprising 35 % by weight of St, 15 % by weight of BA, 1.0 % by weight of GMA, 1.0 part of 2EHTG and 0.7 part of TBHP. One hour after completion of the addition, to this was further continuously added for thirty minutes a mixture comprising 15 10 % by weight of MMA and 0.1 part of TBHP. After completion of the addition, this was further stirred for one hour, and then cooled to obtain a copolymer in latex.

Polymerization conversion was 99.5 %. The obtained latex was coagulated by an aqueous solution of calcium chloride. After the 20 latex was heated up to 90°C, the latex was filtrated with a centrifugal dehydrator. The obtained dehydrated cake of the copolymer was washed with water and dried for 15 hours at 50°C with a parallel-flow dryer to obtain a white powdery polymer sample (11) prepared by three steps polymerization. A weight average molecular weight of the 25 obtained sample was 80,000 measured by GPC. The obtained sample was subjected to the above-mentioned evaluation on the roll peeling property. Table 2 shows the results.

EXAMPLES 9 to 14 and COMPARATIVE EXAMPLES 4 to 6

According to compositions shown in Table 2, samples (12) to (20) were obtained by the same manner as in Example 8. The obtained samples were subjected to the above-mentioned evaluation on the roll peeling property. Table 2 shows the results.

As indicated by samples (11) and (14) to (19), from the results shown in Table 2, it is found that a composition having an excellent roll peeling property is obtained when the composition of the monomer mixture is in the range of the present invention. On the other hand, there is a reduction in the roll peeling property when the composition is out of the range of the present invention, that is, in the case of samples (12), (13) and (20).

TABLE 2

Example No.	Sample	Composition of mixture before polymerization (% by weight)											Chain transfer agent (parts)	Organic peroxide (parts)	Roll peeling property	Weight average molecular weight
		Polymer at first step			Polymer at second step				Polymer at third step							
		MMA	AN	St	BA	EA	2EHA	St	GMA	MMA	AN	2EHG				
Ex. 8	(11)	24		15	15			35	1.0	10		1.0	1.0		10	80,000
Com. Ex. 4	(12)	24		16	15			35	0	10		1.0	1.0		4	80,000
Com. Ex. 5	(13)	21		14	13			31	12	9		1.0	1.0		4	90,000
Ex. 9	(14)	6		4	83			6	1.0	3		1.0	1.0		9	80,000
Ex. 10	(15)	23		15	11			39	1.0	12		1.0	1.0		9	80,000
Ex. 11	(16)	24		15		15		35	1.0	10		1.0	1.0		9	80,000
Ex. 12	(17)	24		15			15	35	1.0	10		1.0	1.0		9	80,000
Ex. 13	(18)		24	15	15			35	1.0		10	1.0	1.0		9	90,000
Ex. 14	(19)	39			50				1.0	10		1.0	1.0		8	80,000
Com. Ex. 6	(20)	24		15	9			41	1.0	10		1.0	1.0		3	80,000

EXAMPLES 15 to 22

The samples (21) to (28) were obtained by setting all the monomers except the amount of GMA used in Example 8 was charged to 100 parts, and adding respective monomers in amounts (parts) shown in  
5 Table 3 in the same manner as in Example 8. The obtained samples were subjected to the above-mentioned evaluation on the roll peeling property. Table 3 shows the results.

As indicated by samples (11) and (21) to (28), from the results shown in Table 3, it is found that a composition having an excellent roll  
10 peeling property is obtained when the kind and amount of (meth)acrylate having oxygen atoms in addition to an ester bond are in the range of the present invention.

TABLE 3

Example No.	Sample	Monomer		Roll peeling property	Weight average molecular weight
		Kind	Amount (parts)		
Ex. 8	(11)	GMA	1.0	10	80,000
Ex. 15	(21)	2HEMA	1.0	10	90,000
Ex. 16	(22)	ETOMA	1.0	10	80,000
Ex. 17	(23)	GA	1.0	10	80,000
Ex. 18	(24)	2HEA	1.0	10	80,000
Ex. 19	(25)	ETOA	1.0	10	90,000
Ex. 20	(26)	{ GMA 2HEA	{ 0.5 0.5	10	80,000
Ex. 21	(27)	GMA	0.3	8	80,000
Ex. 22	(28)	GMA	3.0	9	80,000
Com. Ex. 4	(12)	GMA	0	4	80,000
Com. Ex. 5	(13)	GMA	12	4	90,000

EXAMPLES 23 to 26 and COMPARATIVE EXAMPLES 7 to 11

The samples (29) to (37) were obtained in the same manner as in Example 8 except that 1.0 part of 2EHTG was replaced by compounds shown in Table 4. The obtained samples were subjected to the above-mentioned evaluation on the roll peeling property. Table 4 shows the results.

As indicated by samples (11), (29), (30), (32) and (33), from the results shown in Table 4, it is found that a composition having an excellent roll peeling property can be obtained when the particular chain

transfer agent is used in an amount of the present invention. On the other hand, roll peeling property becomes low when the particular chain transfer agent is not used or an amount of the chain transfer agent is out of the range of the present invention, that is, in the case of samples (31) and (34) to (37).

TABLE 4

Example No.	Sample	Chain transfer agent		Roll peeling property	Weight average molecular weight
		Kind	Amount (parts)		
Ex. 8	(11)	2EHTG	1.0	10	80,000
Ex. 23	(29)	2EHTG	0.5	9	130,000
Ex. 24	(30)	2EHTG	3.0	9	60,000
Com. Ex. 7	(31)	2EHTG	0	4	210,000
Ex. 25	(32)	2-ethylhexyl-β-mercaptopropionate	1.0	10	80,000
Ex. 26	(33)	{ 2EHTG	0.5	10	80,000
		{ 2-ethylhexyl-β-mercaptopropionate	0.5		
Com. Ex. 8	(34)	tert-dodecyl mercaptane	1.0	5	80,000
Com. Ex. 9	(35)	n-dodecyl mercaptane	1.0	5	80,000
Com. Ex. 10	(36)	n-octyl mercaptane	1.0	5	80,000
Com. Ex. 11	(37)	thioglycolic acid	1.0	4	80,000

EXAMPLES 27 to 29 and COMPARATIVE EXAMPLES 12 to 15

The samples (38) to (44) were obtained in the same manner as in Example 8 except that 1.0 part of TBHP was replaced by compounds shown in Table 5. The obtained samples were subjected to the above-mentioned evaluation on the roll peeling property. Table 5 shows the results.

As indicated by samples (11), (38) to (40), from the results shown in Table 5, it is found that a composition having an excellent roll peeling property is obtained when the particular organic peroxide having a tertiary butylperoxy group is used in an amount of the present invention.

On the other hand, the roll peeling property becomes low, when the particular organic peroxide having a tertiary-butylperoxy group is not used or an amount thereof is out of the range of the present invention, that is, in the case of samples (41) to (44).

TABLE 5

Example No.	Sample	Organic peroxide		Roll peeling property	Weight average molecular weight
		Kind	Amount (parts)		
Ex. 8	(11)	TBHP	1.0	10	80,000
Ex. 27	(38)	TBHP	0.3	9	180,000
Ex. 28	(39)	di-tert-butyl peroxide	1.0	10	80,000
Ex. 29	(40)	{TBHP	0.5	10	80,000
		{di-tert-butyl peroxide	0.5		
Com. Ex. 12	(41)	paramentane hydroperoxide	1.0	6	80,000
Com. Ex. 13	(42)	cumene hydroperoxide	1.0	5	80,000
Com. Ex. 14	(43)	benzoyl peroxide	1.0	5	80,000
Com. Ex. 15	(44)	{TBHP	0.05	6	80,000
		{cumene hydroperoxide	1.0		

EXAMPLES 30 to 32 and COMPARATIVE EXAMPLES 16 and 17

Sample (11) obtained in Example 8 was blended with vinyl chloride resin and additives at the ratios shown in Table 6, and the obtained samples were subjected to the transparency evaluation and the evaluation on the roll peeling property. Table 6 shows the results.

From the results shown in Table 6, it is found that a composition having an excellent roll peeling property is obtained when the amounts of addition are in the range of the present invention.

TABLE 6

Example No.	Poly(vinyl chloride) (parts)	Polymer (11) (parts)	Stabilizer (parts)	Lubricant (parts)	DOP (parts)	Transparency		Roll peeling property
						Transmittance (%)	Haze (%)	
Ex. 30	100	1.0	2.0	1.0	3.0	84.2	20.8	10
Ex. 31	100	0.3	2.0	1.0	3.0	84.8	20.8	7
Ex. 32	100	15	2.0	1.0	3.0	80.8	21.2	8
Com. Ex. 16	100	0	2.0	1.0	3.0	86.0	20.6	1
Com. Ex. 17	100	25	2.0	1.0	3.0	73.7	28.1	3

EXAMPLES 33 to 40 and COMPARATIVE EXAMPLES 18 to 25

Sample (11) obtained in Example 8 was blended with a thermoplastic resin and a mixture of poly(vinyl chloride) and another thermoplastic resin at the ratios shown in Table 7, and the obtained  
5 samples were subjected to the above-mentioned evaluation on the roll peeling property. Table 7 shows the results.

From the results shown in Table 7, it is found that an excellent roll peeling property is obtained, when the amounts of addition are in the range of the present invention. In Examples 30, 33, 40 and  
10 Comparative Examples 16, 18 and 25, there were added 2.0 parts of octyl tin mercaptan based stabilizer (TVS #8831 available from Nitto Kasei K.K.), 1.0 part of a lubricant (KALCOHL 8668 available from Kao Corp.), and 3.0 parts of DOP thereto.

In the Table, PVC represents poly(vinyl chloride), CPVC  
15 represents chlorinated poly(vinyl chloride), PP represents polypropylene, PET represents poly(ethylene terephthalate), PC represents polycarbonate, PS represents polystyrene, PMMA represents poly(methyl methacrylate) and ABS represents an ABS resin.

TABLE 7

Example No.	Poly(vinyl chloride) resins (parts)			Thermoplastic resin (parts)						Polymer (11) (parts)	Roll peeling property
	PVC	CPVC	PP	PET	PC	PS	PMMA	ABS			
Ex. 30	100								1.0	10	
Ex. 33		100							1.0	9	
Ex. 34			100						1.0	9	
Ex. 35				100					1.0	9	
Ex. 36					100				1.0	9	
Ex. 37						100			1.0	9	
Ex. 38							100		1.0	9	
Ex. 39								100	1.0	10	
Ex. 40	70							30	1.0	10	
Com. Ex. 16	100								0	1	
Com. Ex. 18		100							0	1	
Com. Ex. 19			100						0	1	
Com. Ex. 20				100					0	1	
Com. Ex. 21					100				0	1	
Com. Ex. 22						100			0	1	
Com. Ex. 23							100		0	1	
Com. Ex. 24								100	0	1	
Com. Ex. 25	70							30	0	1	

#### EXAMPLE 41

The sample (45) was obtained in the same manner as in Example 1 except that 1.0 part of 2EHTG was replaced by 1.0 part of tertiary-dodecyl mercaptan (TDM). Polymerization conversion was 99.6 %. A weight average molecular weight of the obtained sample was 70,000 measured by GPC.

The obtained sample was subjected to the above-mentioned evaluation on the roll peeling property. Table 8 shows the results.

#### EXAMPLES 42 to 49 and COMPARATIVE EXAMPLES 26 to 29

According to compositions shown in Table 8, samples (46) to (57) were obtained in the same manner as in Example 41.

The obtained samples were subjected to the above-mentioned evaluation of the roll peeling property. Table 8 shows the results.

As indicated by samples (45) to (47) and (51) to (56), from the results shown in Table 8, it is found that a composition having an excellent roll peeling property is obtained when the composition of the monomer mixture is in the range of the present invention. On the other hand, when the composition is out of the range of the present invention, that is, in the case of samples (48) to (50) and (57), there is a decline of the roll peeling property.

TABLE 8

Example No.	Sample	Composition of mixture before polymerization (% by weight)										Chain transfer agent (parts)	Organic peroxide (parts)	Roll peeling property	Weight average molecular weight
		Polymer at first step					Polymer at second step								
		MMA	AN	St	BA	EA	2EHA	St	GMA	TDM	TBHP				
Ex 41	(45)	30		19	15			35	10	1.0	1.0			10	70,000
Ex 42	(46)	30		19	15			35	10	0.2	1.0			9	150,000
Ex 43	(47)	30		19	15			35	10	0	1.0			7	210,000
Com. Ex 26	(48)	30		20	15			35	0	1.0	1.0			4	70,000
Com. Ex 27	(49)	30		20	15			35	0	0	1.0			3	210,000
Com. Ex 28	(50)	26		18	13			31	12	1.0	1.0			4	70,000
Ex 44	(51)	6		4	83			6	10	1.0	1.0			8	70,000
Ex 45	(52)	30		19	11			39	10	1.0	1.0			8	70,000
Ex 46	(53)	30		19		15		35	10	1.0	1.0			8	70,000
Ex 47	(54)	30		19			15	35	10	1.0	1.0			8	80,000
Ex 48	(55)		30	19	15			35	10	1.0	1.0			8	80,000
Ex 49	(56)	49			50				10	1.0	1.0			8	80,000
Com. Ex 29	(57)	30		19	8			42	10	1.0	1.0			2	70,000

#### EXAMPLE 50

The same manner as in Example 1 was used except that 1.0 part of 2EHTG was replaced by 1.0 part of TDM to obtain sample (58). The polymerization conversion was 99.7 %. The weight average  
5 molecular weight of the resulting sample was measured by GPC, thereby obtaining 90,000.

The resulting sample was subjected to the above-mentioned evaluation on the roll peeling property. Table 9 shows the results of evaluation.

10

#### EXAMPLES 51 to 58 and COMPARATIVE EXAMPLES 30 to 33

According to compositions shown in Table 9, samples (59) to (70) were obtained in the same manner as in Example 50. The obtained samples were subjected to the above-mentioned evaluation on the roll  
15 peeling property. Table 9 shows the results.

As indicated by samples (58) to (60) and (64) to (69), from the results shown in Table 9, it is found that a composition having an excellent roll peeling property is obtained when the composition of the monomer mixture is in the range of the present invention. On the other  
20 hand, when the composition is out of the range of the present invention, that is, in the case of samples (61) to (63) and (70), there is a decline of the roll peeling property.

TABLE 9

Example No.	Sample	Composition of mixture before polymerization (% by weight)										Chain transfer agent (parts)	Organic peroxide (parts)	Roll peeling property	Weight average molecular weight		
		Polymer at first step		Polymer at second step						Polymer at third step						TDM	TBHP
MMA	AN	St	EA	2EHA	St	GMA	MMA	AN									
Ex50	(58)	24		15	15		35	1.0	10		1.0	1.0		10	90,000		
Ex51	(59)	24		15	15		35	1.0	10		0.2	1.0		9	150,000		
Ex52	(60)	24		15	15		35	1.0	10		0	1.0		7	200,000		
Com.Ex30	(61)	24		16	15		35	0	10		1.0	1.0		4	90,000		
Com.Ex31	(62)	24		16	15		35	0	10		0	1.0		3	210,000		
Com.Ex32	(63)	21		14	13		31	12	9		1.0	1.0		4	80,000		
Ex53	(64)	6		4	83		6	1.0	3		1.0	1.0		8	80,000		
Ex54	(65)	23		15	11		39	1.0	12		1.0	1.0		9	90,000		
Ex55	(66)	24		15		15	35	1.0	10		1.0	1.0		8	90,000		
Ex56	(67)	24		15			35	1.0	10		1.0	1.0		8	80,000		
Ex57	(68)		24	15	15		35	1.0		10	1.0	1.0		8	90,000		
Ex58	(69)	39			50			1.0	10		1.0	1.0		8	80,000		
Com.Ex33	(70)	24		15	8		42	1.0	10		1.0	1.0		2	80,000		

# EXAMPLES 59 to 66

Samples (71) to (78) were obtained in the same manner as in Example 50, by setting all monomers except GMA used in Example 50 to 100 parts by weight, and using respective monomers shown in Table 3 in the respective blending amounts (parts by weight). The obtained samples were subjected to the above-mentioned evaluation of the roll peeling property. Table 10 shows the results.

As indicated by samples (58) and (71) to (78), from the results shown in Table 9, it is found that a composition having an excellent roll peeling property is obtained when the (meth)acrylate having an oxygen atom in addition to an ester bond is used in an amount of the present invention.

TABLE 10

Example No.	Sample	Monomer		Roll peeling property	Weight average molecular weight
		Kind	Amount (parts)		
Ex. 50	(58)	GMA	1.0	10	90,000
Ex. 59	(71)	2HEMA	1.0	10	80,000
Ex. 60	(72)	ETOMA	1.0	10	80,000
Ex. 61	(73)	GA	1.0	10	80,000
Ex. 62	(74)	2HEA	1.0	10	80,000
Ex. 63	(75)	ETOA	1.0	10	90,000
Ex. 64	(76)	{ GMA 2HEA	{ 0.5 0.5	10	90,000
Ex. 65	(77)	GMA	0.3	8	80,000
Ex. 66	(78)	GMA	3.0	9	80,000
Com. Ex. 30	(61)	GMA	0	4	90,000
Com. Ex. 32	(63)	GMA	12	4	80,000

EXAMPLES 67 to 69 and COMPARATIVE EXAMPLES 34 to 37

The samples (79) to (85) were obtained in the same manner as in Example 50 except that 1.0 part of TBHP was replaced by a polymerization initiator shown in Table 11. The obtained samples were subjected to the above-mentioned evaluation on the roll peeling property. Table 11 shows the results.

As indicated by samples (79) to (81), from the results shown in Table 11, it is found that a composition having an excellent roll peeling property is obtained when the particular polymerization initiator

is used in an amount of the present invention. On the other hand, when the particular polymerization initiator is not used and an amount thereof is out of the range of the present invention, that is, in the case of samples (82) to (85), there is a reduction in the roll peeling property.

TABLE 11

Example No.	Sample	Organic peroxide		Roll peeling property	Weight average molecular property
		Kind	Amount (parts)		
Ex. 50	(58)	TBHP	1.0	10	90,000
Ex. 67	(79)	TBHP	0.3	9	150,000
Ex. 68	(80)	di-tert-butyl peroxide	1.0	10	80,000
Ex. 69	(81)	{TBHP	0.5	10	90,000
		{di-tert-butyl peroxide	0.5		
Com. Ex. 34	(82)	paramentane hydroperoxide	1.0	3	80,000
Com. Ex. 35	(83)	cumene hydroperoxide	1.0	3	90,000
Com. Ex. 36	(84)	benzoyl peroxide	1.0	3	90,000
Com. Ex. 37	(85)	{TBHP	0.05	3	90,000
		{cumene hydroperoxide	1.0		

Sample (58) obtained in Example 50 was blended with a vinyl chloride resin and additives at the ratios shown in Table 12. The obtained samples were subjected to the evaluation on the transparency and roll peeling property. Table 12 shows the results.

From the results shown in Table 12, it is found that a composition having an excellent roll peeling property is obtained when the amounts of addition are in the range of the present invention.

TABLE 12

Example No.	Poly(vinyl chloride) (parts)	Polymer (58) (parts)	Stabilizer (parts)	Lubricant (parts)	DOP (parts)	Transparency		Roll peeling property
						Transmittance (%)	Haze (%)	
Ex. 70	100	1.0	2.0	1.0	3.0	84.0	21.0	10
Ex. 71	100	0.3	2.0	1.0	3.0	85.4	20.9	7
Ex. 72	100	15	2.0	1.0	3.0	81.3	21.3	9
Com. Ex. 38	100	0	2.0	1.0	3.0	86.0	20.7	1
Com. Ex. 39	100	25	2.0	1.0	3.0	72.4	29.8	4

EXAMPLES 73 to 80 and COMPARATIVE EXAMPLES 40 to 47

Sample (58) obtained in Example 50 was blended with a thermoplastic resin and a mixture of poly(vinyl chloride) and another thermoplastic resin at the ratios shown in Table 13, and the obtained  
5 samples were subjected to the evaluation on the roll peeling property. Table 13 shows the results.

From the results shown in Table 13, it is found that a composition having an excellent roll peeling property is obtained when the amounts of addition are in the range of the present invention. In  
10 Examples 70, 73, 80 and Comparative Examples 38, 40 and 47, there were added 2.0 parts of octyl tin mercaptan based stabilizer (TVS #8831 available from Nitto Kasei K.K.), 1.0 part of a lubricant (KALCOHL 8668 available from Kao Corp.), and 3.0 parts of DOP thereto.

TABLE 13

Example No.	Poly(vinyl chloride) resins (parts)			Thermoplastic resin (parts)						Polymer (58) (parts)	Roll peeling property
	PVC	CPVC	PP	PET	PC	PS	PMMA	ABS			
Ex. 70	100								1.0	10	
Ex. 73		100							1.0	9	
Ex. 74			100						1.0	9	
Ex. 75				100					1.0	9	
Ex. 76					100				1.0	9	
Ex. 77						100			1.0	9	
Ex. 78							100		1.0	9	
Ex. 79								100	1.0	10	
Ex. 80	70							30	1.0	10	
Com. Ex. 38	100								0	1	
Com. Ex. 39		100							0	1	
Com. Ex. 40			100						0	1	
Com. Ex. 41				100					0	1	
Com. Ex. 42					100				0	1	
Com. Ex. 43						100			0	1	
Com. Ex. 44							100		0	1	
Com. Ex. 45								100	0	1	
Com. Ex. 46	70							30	0	1	

## INDUSTRIAL APPLICABILITY

A thermoplastic resin composition containing the processing aid for thermoplastic resin of the present invention is excellent in peeling property from a metal surface at a high temperature as compared with  
5 conventional resins, and provides a longer processable time.

## CLAIMS

1           1. A processing aid for thermoplastic resin having a weight  
2 average molecular weight of 10,000 to 300,000, which is obtained by  
3 polymerizing an alkyl (meth)acrylate, or an alkyl (meth)acrylate and  
4 another vinyl monomer copolymerizable therewith, in the presence of a  
5 mercaptan having an alkyl ester group with C<sub>4-20</sub> alkyl group as a chain  
6 transfer agent, and/or an organic peroxide having a tertiary-butyl  
7 peroxy group as a polymerization initiator.

1           2. The processing aid for thermoplastic resin of Claim 1,  
2 which is obtained by polymerizing 0.1 to 10 % by weight of the  
3 (meth)acrylate having an oxygen atom in addition to an ester bond as  
4 said alkyl (meth)acrylate, 10 to 99.9 % by weight of another alkyl  
5 acrylate, and 0 to 89.9 % by weight of another vinyl monomer  
6 copolymerizable therewith.

1           3. The processing aid for thermoplastic resin of Claim 1,  
2 wherein an amount of the organic peroxide having a tertiary-butyl  
3 peroxy group is 0.1 to 5 parts by weight based on 100 parts by weight of  
4 the entire monomers.

1           4. A thermoplastic resin composition comprising 100 parts  
2 by weight of a thermoplastic resin and 0.1 to 20 parts by weight of the  
3 processing aid for thermoplastic resin of Claim 1.

1           5. The thermoplastic resin composition of Claim 4, wherein

2 the thermoplastic resin is at least one resin selected from the group  
3 consisting of poly(vinyl chloride) resins, polyethylene, polypropylene,  
4 polyester, polystyrene, polycarbonate, a methacrylate resin and an ABS  
5 resin.

1 6. The processing aid for thermoplastic resin of Claim 2,  
2 wherein the (meth)acrylate having an oxygen atom in addition to an ester  
3 bond is a (meth)acrylate containing an epoxy group.

1 7. The processing aid for thermoplastic resin of Claim 2,  
2 wherein the (meth)acrylate having an oxygen atom in addition to an ester  
3 bond is a (meth)acrylate containing a hydroxy group.

1 8. The processing aid for thermoplastic resin of Claim 2,  
2 wherein the (meth)acrylate having an oxygen atom in addition to an ester  
3 bond is a (meth)acrylate containing an alkoxy group.

[illegible][illegible]

# Declaration For U.S. Patent Application

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention **entitled**

(Insert Title) PROCESSING AID FOR THERMOPLASTIC RESIN AND THERMOPLASTIC RESIN COMPOSITION  
CONTAINING THE SAME

the specification of which is attached hereto unless the following is checked:

was filed on April 14, 2000 / \_\_\_\_\_ as United States Application Number or PCT International  
☒ Application Number PCT/JP00/02435 / \_\_\_\_\_ and was amended on \_\_\_\_\_  
 (if applicable).

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claim(s), as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to patentability as defined in Title 37, Code of Federal Regulations, § 1.56.

I hereby claim foreign priority benefits under Title 35, United States Code, § 119 (a) - (d) of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

(List prior foreign applications. See note A on back of this page)	<u>116791/1999</u>	<u>Japan</u>	<u>23/4/1999</u> /	Priority Claimed
	(Number)	(Country)	(Day/Month/Year Filed)	<input checked="" type="checkbox"/> Yes <input type="checkbox"/> No
	<u>134924/1999</u>	<u>Japan</u>	<u>14/5/1999</u> /	<input checked="" type="checkbox"/> Yes <input type="checkbox"/> No
	(Number)	(Country)	(Day/Month/Year Filed)	<input type="checkbox"/> Yes <input type="checkbox"/> No
	(Number)	(Country)	(Day/Month/Year Filed)	<input type="checkbox"/> Yes <input type="checkbox"/> No

(See note B on back of this page)

☐ See attached list for additional prior foreign applications

I hereby claim the benefit under Title 35, United States Code, § 119(e) of any United States provisional application(s) listed below.

_____ (Application Number)	_____ (Filing Date)
_____ (Application Number)	_____ (Filing Date)

I hereby claim the benefit under Title 35, United States Code, § 120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, § 112, I acknowledge the duty to disclose information which is material to patentability as defined in Title 37, Code of Federal Regulations, § 1.56 which became available between the filing date of the prior application and the national or PCT international filing date of the application:

(List Prior U.S. Applications)	_____ (Application Serial Number)	_____ (Filing Date)	_____ (Status) (patented, pending, abandoned)
	_____ (Application Serial Number)	_____ (Filing Date)	_____ (Status) (patented, pending, abandoned)

I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and to transact all business in the Patent and Trademark Office connected therewith:

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I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Title 18 of the United States Code, § 1001 and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

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